## **Periodic Table** Organizing the Elements into the Periodic Table

Directions: Check off each step as you complete it. Section 5-1 Arranging the Elements

Simply read p.108 – 112 from the textbook (YOU DO NOT NEED TO TAKE PICTURES FOR THIS SECTION) thoughtfully & quietly to yourself.

Section 5-2 Design of the Periodic Table

- Take pictures of pages 112, 113, 116 & 117.
- Read pages 112-117 thoughtfully & quietly to yourself.
- Highlight the information related to the following key terms:
  - o Groups or families
  - o Periods
  - Atomic number (location of on a periodic table square)
  - Atomic mass (location of on a periodic table square)
    - HELPFUL HINT: For now, think of atomic mass & mass # as the same thing

1) Explain why anyone would arrange the elements into the periodic table:

| 2) What are <b>GROUPS/FAMILIES</b> in the periodic table?          |                |  |
|--|----------------|--|
| 3) What are PERIODS in the periodic table?                         | 6 -            |  |
| <ol> <li>Label the 4 parts of the element square below:</li> </ol> | C←             |  |
|  | →Carbon<br>→12 |  |